

## LEAD ACID BATTERIES, INDUSTRIAL WET

# **MATERIAL SAFETY DATA SHEET**

# SECTION: 1 PRODUCT IDENTIFICATION

Chemical/ Trade Name : Lead Acid Battery

Chemical Family/ Classification : Electric Storage Battery :

Manufacture's Name : Sparco Batteries Pvt. Ltd.

Address : Kila no.-22, Khasra no. 23/1/2, 23/2/1,

Vill- Nathupur, Sonipat-131029, India.

Person Responsible for Preparation : Mr. Pramod Kumar For Emergency : Mr. Arvind Tuli

Other Information Calls : +91-9891-702006 (Pramod Kumar )

MSDS Issued Date : March 30th 2018 MSDS Revised Date : NA. First Version

MSDS Revision No. : 0

# SECTION: 2 HAZARDOUS INGREDIENTS / IDENTITY INFORMATION

Principal Components (Chemical & Common Name)		C.A.S. #	% By Weight	Air Exposure Limits (g/m³)		
Inorganic compounds of				OSHA	ACGIH	NIOSH
	Lead	7439-92-1	55-65	50	150	100
	Antimony	7440-36-0	2.0 max	500	500	
	Tin	7440-31-5	0.1 max	10	200	
Sulphuric Acid (Electrolyte)		7464-93-9	20-25	1000	1000	1000
Poly Propylene (Battery Case)		9003-07-0	6-8	NA	NA	NA
Poly Ethylene (Separator)		9002-88-4	1-2	NA	NA	NA

**Note:** Inorganic lead and Electrolyte (Sulphuric Acid) are the primary components of battery manufactured by Sparco Batteries Pvt. Ltd. Other ingredients may be present depending upon battery type. Poly Propylene is the principal case material of all our Batteries. Contact your Sparco Batteries Pvt. Ltd's representative for additional information.



# **SECTION: 3 PHYSICAL AND CHEMICAL PROPERTIES**

Properties	Finished Battery	Lead	Sulphuric Acid	
Appearance	Solid Object	Bluish Gray	Clear Liquid	
		Heavy metal		
Oder	NA	Odorless	Slightly Acid	
Sp. Gravity gm/cc (H <sub>2</sub> O-	NA	11.34	1.215 - 1.250	
1.00)				
Melting Point	NA	327°C	NA	
Boiling Point	NA	1740°C	290°C	
Solubility	NA	Insoluble in	100% miscible in water,	
		water	liberates heat	
Vapour Density (Air - 1)	NA	NA	>1	
Vapour Pressure (mm	NA	NA	11-17	
Hg) @ 20°C				
Evaporation rate (Butyl	NA	NA	NA	
acetate=1)				
% Volatile by weight	NA	NA	NA	

SECTION: 4 STABILITY & REACTIVITY

Stability : Stable

**Conditions to Avoid** : Avoid Prolonged Overcharging & sources of ignitions.

Use only approved charging Methods.

Do not puncture battery case.

Materials to Avoid (Incompatibility)

**Sulphuric acid** : Contact with combustibles and organic materials may cause fire & explosion.

Also reacts violently with reducing agents, metals SO<sub>3</sub> gas, strong oxidizers and water contact with metals may produce toxic SO<sub>3</sub> fumes and may release

flammable hydrogen gas.

Lead Compounds : Avoid contact with strong acids, bases, halides, halogenated Potassium nitrate

permanganate peroxides, nascent hydrogen and reducing agents.



## **Hazardous Decomposition Products**

**Sulphuric Acid** : Sulphuric trioxide, carbon monoxide, sulphuric acid must. Sulphur dioxide and

hydrogen sulphide.

**Lead compounds**: High temperatures likely to produce toxic metal fume, vapour, or dust: contact with

strong acid or base or presence of nascent hydrogen may generate highly arsine gas,

toxic.

Hazardous

**Polymerization**: Will not occur.

#### SECTION: 5 HEALTH HAZARD INFORMATION

Under normal conditions of battery use, internal components will not present a health hazard. The following information is provided for battery electrolyte (sulphuric acid) and lead for exposure that may occur during battery production or container breakage or under extreme heat conditions such as fire.

#### **Routes of Entry:**

Inhalation : Yes
Ingestion : Yes
Eye contact : Yes
Skin Contact : Yes

# Signs and Symptoms of Exposure

#### **Acute Hazard**

Lead:

Direct skin or eye contact may cause local irritation.

Inhalation or ingestion of lead dust or fumes may result in head ache, Vomiting, fatigue, sleep disturbance, weight loose, anemia and leg arm and joint pain.

Sulphuric Acid:

H<sub>2</sub>SO<sub>4</sub> is corrosive and contact may cause skin irritation and burns.

#### **Chronic Health Effects**

Lead:

Prolonged exposure may cause central nervous system damage, gastro intestinal disturbances, anemia, wrist drop and kidney disinfection. Battery posts. Terminals and other related accessories containing lead and lead compound may cause cancer and reproductive harm. Wash hands after handling lead components.

Sulphuric Acid:

Repeated contact with  $H_2SO_4$  cause skin burns, erosion of teeth  $\,$ , irritation or inflammation of nose, throat and lungs.



# SECTION :6 EMERGENCY AND FIRST AID PROCEDURES

# Emergency & First Aid Procedures

#### • Inhalation:

Not expected for product under normal conditions of use, However acid vapours can be released due to overcharging or abuse of battery. In such cases move exposed person to fresh air if he finds inconvenient or becomes fatigue. If needed oxygen may be administered. Seek medical attention.

#### Eyes

Immediately flush with water for at least 15 minutes. If irritation develops, seek prompt medical attention.

#### • Skin

Flush the contacted area with large amount of water and mild soap. If irritation develops, seek medical attention.

#### Ingestion

Do not include vomiting. If conscious drink large amount of water/milk. Obtain medical attention.

NOTE: Never give anything by mouth to an unconscious person.

## SECTION :7 FIRE AND EXPLOSION HAZARD DATA

Flash Point : Not Applicable

Flammable Limits : LEL = 4.1% (Hydrogen gas in air); UEL= 74.2% Extinguishing Media : Multi Purpose Dry Chemical Powder or CO2

Fire fighting Procedure: Extinguish fire with the agent suitable for surrounding

combustible materials. Use Self-contained breathing apparatus to prevent possible inhalation of acid mists, smokes and decomposed products due to fire. Cool the exterior of the battery's to prevent

rupture.

Unusual Fire and Explosion Hazard

Sulphuric Acid vapors are generated upon overcharge and

Polypropylene case may burst in such cases.

Hydrogen gas may be produced during overcharge and may explode

if igniting.

Avoid open flames/sparks/other sources of ignition near battery and provide good ventilation to prevent the above two

incidents.



SECTION: 8 CONTROL MEASURES

Engineering Controls : Store and handle in well-ventilated area. If mechanical ventilation is

used, components must be acid resistant.

Work Practices : Handle batteries cautiously to avoid spills, Make sure that vent caps

are made tight. Avoid contacts with internal components.

Personal Precautions : Acid Resistant aprons, boots and protective clothing, rubber or plastic

acid resistant gloves with elbow length gauntlet. ANSI approved safety

glasses with side shields face shield recommended.

Environmental : Lead and its compounds can pose a serve threat to the environment.

Precautions Contamination of water, soil and air should be prevented.

Emergency flushing : In areas where sulphuric acid is handled in concentrations greater than 1%,

emergency eyewash stations and showers should be provided with unlimited

water supply.

## SECTION: 9 PRECAUTIONS FOR SAFE HANDLING AND STORAGE

Spill or Leak procedures : • Stop flow of material, if possible carefully neutralize spilled

electrolyte with soda ash, sodium bicarbonate or lime.

• Dispose of contaminated material in accordance with applicable

local, state and federal regulations.

• Do not release un-neutralized acid.

Handling : • Do not carry batteries by terminal.

• Do not drop battery, attempt to open or puncture battery case.

Avoid flames and sparks nearby during and immediately after

charging of battery.

Storage : • Store batteries in cool, dry well-ventilated areas.

• Batteries should be stored under roof for protection against adverse

weather conditions.

Hygiene : • Wash hands thoroughly before eating or drinking after handling batteries.

• Work clothes and equipment should remain in designated

lead contaminated areas, and never taken home or

laundered with personal clothing.



SECTION: 10 TOXICOLOGICAL INFORMATION

General : The primary routes and exposure to lead are ingestion or inhalation of lead dust and fumes.

But this is not applicable for finished product under normal conditions of use.

Acute : Exposure to lead and its compounds may cause Head Ache, Vomiting, Fatigue, Sleep

Disturbances, Weight Loss, Kidney Damage, Anemia and Pain in Legs, Arms and Joints.

Chronic : Exposure to lead and its compounds may cause Central Nervous System damage,

Gastrointestinal disturbances, Anemia, Wrist drop etc.

SECTION: 11 DISPOSAL CONSIDERATIONS

• Lead Acid batteries are completely recyclable. Return the whole scrap battery(s) to the distributor, manufacturer or authorized lead smelter for recycling.

- For neutralized spills, place the residue in an acid resistant container with sorbent material like sand and dispose off in accordance with local, state and federal regulations for acid and lead compounds.
- Contact local and / or state environmental officials for a clear disposal information.

#### SECTION: 12 TRANSPORT INFORMATION

IMDG Proper Shipping Name : Batteries, Wet, Filled with Acid

IMDG UN Class : 8

IMDG UN Identification : UN 2794 IMDG Label : None Record

IMDG Packing Group : III

IATA Proper Shipping Name : Batteries, Wet, Filled with Acid

IATA UN Class : 8

IATA UN Identification : UN 2794 IATA Label : Corrosive



SECTION: 13 REGULATORY INFORMATION

Hazardous under Hazard : Lead : Yes Communication Standards : Sulphuric Acid : Yes

For Hazard Category & Exposure Limits, refer Secton - 2

SECTION: 14 OTHER INFORMATION

The information and recommendations contained herein have been compiled from sources believed to be reliable and to represent current knowledge on the subject Sparco Batteries Pvt. Ltd. Its subsidiaries or affiliates assume no responsibility in connection therewith, nor can be assumed that all acceptable safety measures are contained herein, or that other additional measures may not be required under particular or exceptional conditions or circumstances.